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Synthesis of Acyl Alkenylindium Reagents and Their Application in the Synthesis of (Z) - α , β -Unsaturated Ketones via Palladium-Catalyzed Cross-Coupling Reaction

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S Supporting Information

reaction of allenyl ketones with indium and indium chloride in methanol under mild conditions. Their synthetic applications were demonstrated from Pd-catalyzed cross-coupling reactions with aryl bromides and iodides and alkenyl and aryl triflates for the synthesis of (Z) - α , β -unsaturated ketones.

C ross-coupling reaction using a variety of organometallic
reagents in the presence of a transition metal catalyst has
have an
a of the mest similar to research subjects for the been one of the most significant research subjects for the formation of C−C and C−heteroatom bonds.¹ In particular, the Pd-catalyzed cross-coupling reaction has been widely used as a pow[e](#page-3-0)rful tool in synthetic chemistry.² To date, development of new catalysts, organometallic reagents, and electrophilic coupling partners has been contin[uo](#page-3-0)usly reported to expand the scope of cross-coupling reactions. Among them, because organoindium reagents showed advantageous properties related to their selectivity and reactivity, ease of handling and preparation, thermal stability, and low toxicity, δ cross-coupling reactions using organoindium reagents have also been of great interest. After Pd-catalyzed cross-coupling [re](#page-3-0)actions using tri(organo)indium reagents were reported by Sarandeses and $\text{co-workers},^4$ coupling reactions using a large number of $organoidium reagents⁵ such as allylindiums, allenylindiums,$ alkenylindi[um](#page-3-0)s, 1,3-butadien-2-ylindiums, tri(naphthyl)indiums, tetra(organo)indates, i[nd](#page-3-0)ium tri(organothiolates), acylindiums, $β$ -phosphoryl alkylindiums, arylindiums, benzylindiums, and alkylindiums 8 have been demonstrated.⁹ Recently, Loh and coworkers reported the synthetic method o[f](#page-3-0) indium homoen[o](#page-3-0)lates *via* oxidative [a](#page-3-0)[d](#page-3-0)dition of indium and indium trichloride to α , β enones¹⁰ and insertion of indium into a β -halo ester.¹¹ In our continuing efforts to develop Pd-catalyzed cross-coupling reactio[ns](#page-3-0) using organoindium reagents, we envision[ed](#page-3-0) that if allenyl ketones would be employed in the reaction with In and/or $InCl_n$ ($n = 1$ and 3), the hydroindation reaction would take place to produce acyl alkenylindium reagents, which can be applicable in Pd-catalyzed cross-coupling reactions, producing α , β -unsaturated ketones. Herein, we report a novel synthetic method of acyl alkenylindium reagents from allenyl ketones and indium and

indium chloride and their synthetic application in Pd-catalyzed cross-coupling reactions for the synthesis of α , β -unsaturated ketones (Scheme 1).

Scheme 1. Synthesis of Acyl Alkenylindium Reagents and Their Synthetic Application in Pd-Catalyzed Cross-Coupling Reaction

At the outset, 1-phenylbuta-2,3-dien-1-one (1a) was selected as the substrate to prepare the organoindium reagent. Investigation of a wide range of indium reagents such as In, InCl, InCl₃, In/InCl, and In/InCl₃ revealed that In and InCl were the reagents of choice for hydroindation (Table 1). Screening of solvents revealed that methanol was an optimal solvent, but other solvents such as CH_3CN/H_2O , THF/H₂O, and MeOH/H₂O gave inferior results. The best result was obtained from a reaction of 1a (0.2 mmol, 1.0 equiv) with In (1.0 equiv) and InCl (0.8 equiv) in MeOH at 30 °C for 30 min, affording benzoyl alkenylindium reagent (2a) in 68% isolated yield together with 1 phenylbut-3-en-1-one (3a) in 19% yield (entry 8).

X-ray diffraction analysis proved that 2a is a benzoylsubstituted alkenylindium, having a monomeric structure with chelation of the two carbonyl groups to indium (Figure 1). The coordination conformation of the central indium metal can be

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^aReactions were carried out with 1a (0.2 mmol, 1 equiv) in solvent (0.8 mL) at 30 °C. ^bNMR yield using CH_2Br_2 as an internal standard.
Closed vield of 2a Isolated yield of 2a.

Figure 1. X-ray structure of benzoyl alkenylindium (2a).

depicted as a distorted trigonal bipyramid, in which C(9), C(19), and $Cl(1)$ occupy the equatorial plane and $O(1)$ and $O(2)$ occupy the apical positions. The average $C=O$ bond length is 1.247 Å, and the average C=C bond $[C(8)-C(9)$ and C(18)– C(19)] lengths, which are adjacent to the In−C bonds, are 1.338 Å.

Next, a wide range of allenyl ketones were studied to demonstrate the scope and limitation of the present method for the synthesis of acyl alkenylindium reagents under optimal reaction conditions (Scheme 2). Electronic variation of substituents on the aryl ring of 1-arylbuta-2,3-dien-1-one (1) slightly affected the reaction efficiency. For example, acyl alkenylindium reagents (2b−d) with electron-donating 2-, 3-, or 4-methyl substituents on the aryl ring were obtained in moderate to good yields ranging from 60 to 70%. 1-(4- Methoxyphenyl)buta-2,3-dien-1-one (1e) bearing a strong electron-donating methoxy group is less reactive, leading to the formation of 4-methoxybenzoyl alkenylindium reagent 2e in 50% yield. Likewise, 2-chlorobenzoyl alkenylindium reagent 2f was prepared from the hydroindation reaction of 1-(2 chlorophenyl)buta-2,3-dien-1-one (1f) with In and InCl in MeOH. Hydroindation of 1-(thiophen-2-yl)buta-2,3-dien-1-one (1g) provided 2g in 60% yield. When alkyl allenyl ketones were subjected to the hydroindation reaction, the product yields slightly decreased due to the acidic α -proton of the carbonyl group. 1-Phenylpenta-3,4-dien-2-one (1h) and 1-phenylhexa-4,5-dien-3-one (1i) were converted to the corresponding alkanoyl alkenylindium reagents 2h and 2i in 40 and 48% yields, respectively. The corresponding allyl ketones were produced in about 10% yields for all of the allenyl ketones. 1-Phenylpenta-2,3-

Scheme 2. Scope of Allenyl Ketones^a

a
Reactions were carried out using In (0.2 mmol, 1.0 equiv), InCl (0.8 equiv), and allenyl ketone 1 (1 equiv) in MeOH (0.8 mL) at 30 °C. All yields are isolated yields. ^b For 30 min. ^cIn (0.2 mmol, 1.0 equiv) and InCl (1.6 equiv) were used. $\frac{d}{dr}$ for 10 min. $\frac{e}{r}$ for 20 min. $\frac{f}{r}$ At 40 $\frac{e}{r}$.

SEq. 5 min. $\frac{h}{r}$ [0.2 mmol 1.0 equiv) and InCl (1.2 equiv) were used. For 5 min. h In (0.2 mmol, 1.0 equiv) and InCl (1.2 equiv) were used.

dien-1-one having a methyl group on the terminal $sp²$ carbon did not undergo the hydroindation reaction.

We next examined Pd-catalyzed cross-coupling of benzoyl alkenylindium reagent 2a with ethyl 4-iodobenzoate (5j) (see Supporting Information). A wide range of solvents such as DMA, THF, toluene, DMSO, and DMF in the presence of $PdCl₂(PPh₃)₂$ (5.0 mol %) were examined in the cross-coupling reaction, and it was found that DMF was the solvent that provided optimal yields. It is noteworthy that the E-product was selectively produced in DMA, while the Z-product was selectively produced in DMSO. A large number of Pd catalysts, such as $Pd(OAc)_2$, $Pd_2(dba)_3$, $Pd(PPh_3)_4$, $Pd(PhCN)_2Cl_2$, $[Pd(\pi-ally)]$ - Cl_2 , and PdCl₂(PPh₃)₂, were screened in DMF. The optimal reaction conditions were obtained from reactions of 2a (0.7 equiv) with 5j (0.15 mmol, 1 equiv) in the presence of $PdCl_2(PPh_3)$ ₂ (5.0 mol %) and LiCl (2.0 equiv) in DMF (1.0 mL) at 100 °C for 2 h, leading to the formation of α , β unsaturated ketone 6j in 86% yield $(Z/E = 1.4:1)$. When LiCl was not used, the product yield slightly decreased 72% and Zselectivity significantly increased $(Z/E = 13:1)$.

To demonstrate the scope and limitation of this cross-coupling reaction, we investigated the reaction of 2a with a large number of aryl iodides and bromides under the optimal reaction conditions (Scheme 3). Electronic modification of substituents on the aryl ring of 2a did not largely influence the reaction efficiency. 1[-Iodobenze](#page-2-0)nes with electron-donating 2-methyl and 4-methoxy substituents underwent the cross-coupling reaction, furnishing the corresponding α , β -unsaturated ketones 6b (60%) and 6c (72%). The cross-coupling reaction was compatible with a fluoro group. 3-Trifluoromethylphenyl iodide 5f was coupled with indium reagent to the desired product 6f. Iodobenzene having electron-withdrawing groups such as formyl, acetyl, ethoxycarbonyl, and nitrile are suitable for the cross-coupling reactions, providing the desired α , β -unsaturated ketones (6g–k) in good yields, ranging from 68 to 86%. There are no products arising from the addition of the nucleophilic acyl alkenylindium

^aReactions were carried out with $2a$ (0.7 equiv), 5 (0.2 mmol, 1 equiv), and LiCl (2.0 equiv) in DMF (1.3 mL) at 100 °C for 2 h. Isolated yield. Numbers in parentheses indicate Z/E ratio. Aryl bromides were used.

reagents to the carbonyls. 1-Iodo-3,5-dimethylbenzene (5l) was subjected to the cross-coupling reaction to deliver 6l in 70% yield. When 1-naphthyl iodide was treated with indium reagent 2a, the cross-coupling reaction took place to produce 6m in 61% yield. Treatment of 3,4-dichloro-1-iodobenzene with 2a afforded the $α, β$ -unsaturated ketone 6n in 85% yield. 2-Iodothiophene was applied to the present Pd-catalyzed coupling reaction, affording 6o in 60% yield. Aryl bromides are less reactive than the corresponding iodides. For instance, 1-bromo-4-chlorobenzene gave the desired product 6e in 51% yield. Ethyl 4-bromobenzoate was also reacted with indium reagent 2a, producing 6j in 61% yield. (Z)- α , β -Unsaturated ketones were produced as the major isomer in all of the cross-coupling reactions.

Next, a large number of alkenyl and aryl triflates were employed in the Pd-catalyzed cross-coupling reactions with 2a (Scheme 4). 4-tert-Butyl cyclohexen-1-yl triflate was applied to

 a Reactions were carried out with 2a (0.7 equiv), 7 (0.2 mmol, 1 equiv), and LiCl (2.0 equiv) in DMF (1.3 mL) at 100 °C for 2 h. Isolated yield. Numbers in parentheses indicate Z/E ratio.

the present coupling reaction, affording the corresponding dienyl ketone (6p) in 70% yield $(Z/E = 12:1)$. When phenyl trifluoromethanesulfonates with 4-chloro and 3,5-dimethyl groups were employed, the desired α , β -unsaturated ketones 6e and 6l were obtained in 82 and 76% yields, respectively. 1- Naphthyl triflate (7m) was compatible with the coupling reaction conditions.

With these results in hand, we scrutinized the cross-coupling reaction of a wide range of acyl alkenylindium reagents 2 with 5j (Scheme 5). Acyl dialkenylindium reagents (2b−d) bearing

Scheme 5. Scope of Acyl Alkenylindium Reagents in the Coupling Reaction α

^aReactions were carried out with 2 (0.7 equiv), 5j (0.2 mmol, 1 equiv), and LiCl (2.0 equiv) in DMF (1.3 mL) at 100 °C for 2 h. Isolated yield. Numbers in parentheses indicate Z/E ratio.

electron-donating 2-, 3-, or 4-methyl substituents on the aryl ring were reacted with 5j to produce the corresponding $\alpha_i\beta$ unsaturated ketones 8a, 8b, and 8c in good yields ranging from 72 to 81%. Indium reagent 2e obtained from 1-(4 methoxyphenyl)buta-2,3-dien-1-one having a strong electrondonating methoxy group also worked, leading to the α , β unsaturated ketone 8d in 78% yield $(Z/E = 25:1)$. 2-Chlorobenzoyl alkenylindium reagent 2f prepared from 1-(2 chlorophenyl)buta-2,3-dien-1-one underwent the cross-coupling reaction with 5j, resulting in the formation of 8e in 73% yield. When acyl alkenylindium 2g with a thiophen-2-yl moiety was employed, the desired α , β -unsaturated ketone 8f was obtained in 82% yield $(Z/E = 26:1)$. The cross-coupling reaction was amenable to reactions with indium reagents generated from 1 phenylpenta-3,4-dien-2-one and 1-phenylhexa-4,5-dien-3-one to afford 8g (62%) and 8h (84%), keeping the high Z-selectivity.

In conclusion, we have developed a synthetic method for the preparation of acyl alkenylindium reagents from the hydroindation reaction of allenyl ketones with indium and indium chloride in methanol under mild conditions. Their synthetic applications were demonstrated from Pd-catalyzed crosscoupling reactions with aryl bromides and iodides and alkenyl and aryl triflates for the synthesis of (Z) - α , β -unsaturated ketones.

■ ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b02043. Experimental procedures, characterization data, X-ray crystallography data (2a), and copies of NMR spectra for all products (PDF)

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Notes

The authors declare no competing financial interest.

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